

Building a Thermodynamic Database for Platinum-Based Superalloys: Part III

RESULTS OF CALCULATIONS FROM FIRST PRINCIPLES

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Work is being done at Mintek, the University of Leeds and the University of Bayreuth to build up a platinum-aluminium-chromium-ruthenium (Pt-Al-Cr-Ru) database for the prediction of phase diagrams for further alloy development by obtaining good thermodynamic descriptions of all of the possible phases in the system. Binary descriptions were combined, allowing extrapolation into the ternary systems, and experimental phase equilibrium data were compared with calculated results. Part I of this series of papers (1) addressed the Pt-Al-Ru system, and Part II (2) the Pt-Cr-Ru system. This final paper (Part III) deals with progress towards a platinum-aluminium-chromium-nickel (Pt-Al-Cr-Ni) database at the University of Bayreuth, using thermodynamic calculations from first principles to deal with the problem of sparse data. The Pt-Al-Cr-Ru and Pt-Al-Cr-Ni databases will eventually be merged to give a Pt-Al-Cr-Ni-Ru database.

Work has been ongoing in building a thermodynamic database for the prediction of phase equilibria in Pt-based superalloys (1–7). The alloys are being developed for high-temperature applications in aggressive environments. The database will aid the design of alloys by enabling the calculation of the composition and proportions of phases present in alloys of different compositions. This paper is a revised account of work presented at the conference: Southern African Institute of Mining and Metallurgy ‘Platinum Surges Ahead’ at Sun City, South Africa, from 8th to 12th October 2006 (7).

Part I, describing initial results for the Pt-Al-Ru

system from the compound energy formalism model, was published in the July 2007 issue of *Platinum Metals Review* (1). Part II (2) described the different approach which was needed for the other binary and ternary combinations within the Pt-Al-Cr-Ru system, with simpler representation to allow for sparse data. This paper (Part III) completes the series by outlining work at the University of Bayreuth on the platinum-aluminium-chromium-nickel (Pt-Al-Cr-Ni) database, using thermodynamic calculations from first principles (*ab initio*). The Pt-Al-Cr-Ni database is eventually to be merged with the Pt-Al-Cr-Ru database to give a Pt-Al-Cr-Ni-Ru database.

Calculations from First Principles (*Ab Initio*)

Part II of this series of papers (2) described a simplified approach to the calculation of phase diagrams, for cases where use of the compound energy formalism model (1) is inappropriate due to the sparseness of experimental data.

At the University of Bayreuth, a different approach has been used to compensate for sparse data, using *ab initio* calculations to determine the enthalpy of formation of intermetallic compounds. These calculations, which are complex and time consuming, are based on density functional theory, as described by Kohn and Sham (8). The power of the technique derives from the fact that the thermodynamic results are often difficult to determine experimentally. The results can be used directly in a thermodynamic description of the alloy systems (9). The VASP program (10) has been used here to calculate the enthalpies of formation of the ordered compounds. At this stage, no magnetic contribution has been incorporated in the calculations, since there is some discrepancy in the experimental observations: for the L_{12} -ordered CrPt_3 Kussmann *et al.* (11) observed ferromagnetic behaviour, whereas Pickard *et al.* (12) observed ferrimagnetic behaviour. The total enthalpies of the pure elements were compared with those obtained by Wang *et al.* (13), with very good agreement.

Resulting Chromium-Platinum Phase Diagram

The calculated phase diagram after thermodynamic optimisation is given in Figure 1 (7), and shows very good agreement with experimental data. The modelled eutectic temperatures agree, within the stated experimental errors, with the results of Venkatraman and Neumann (14) (derived on the basis of Massalski (15)), rather than those of Oikawa *et al.* (16). The ordering reaction still needs to be modelled correctly, and a later version of the Cr-Pt phase diagram shows promise (17). The small phase region of the L_{12} CrPt_3 phase at low temperatures may be a result of not yet having taken the magnetic properties into account. Since all the Gibbs energies have now been modelled, it is possible to calculate further thermodynamic data. Figure 2 shows a comparison between the calculated and experimental chemical activities (18, 20). Using *ab initio* calculations in the thermodynamic model has predicted the presence of L_{12} Cr_3Pt at low temperatures. This suggests that the platinum-rich side of the phase diagram should be examined further, probably using the diagram from Zhao (17).

Platinum-Aluminium-Chromium-Nickel

Thermodynamic assessment of the Cr-Pt system has already shown that many regions of the binary

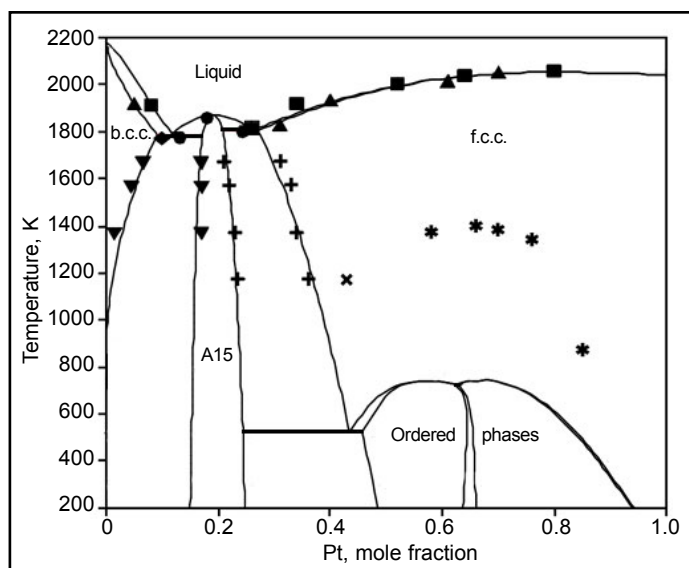


Fig. 1 Cr-Pt phase diagram calculated using *ab initio* values. Symbols denote experimental values from Reference (14) and references therein

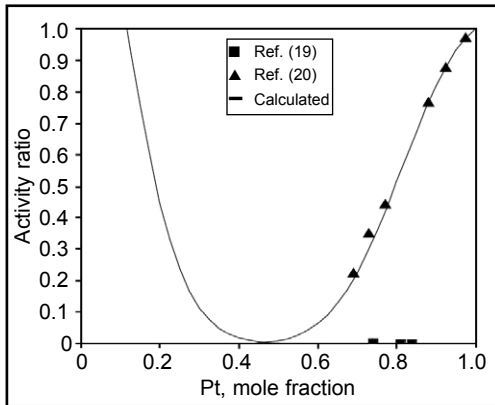


Fig. 2 Calculated activity of Cr and Pt at 1500°C (with respect to the pure phases at 1500°C) compared with experimental results (19, 20)

phase diagram are relatively unknown. The *ab initio* calculations used here show that a stable L_{12} Cr_3Pt structure is achieved. Greenfield and Beck (21) discovered a stable L_{12} structure at 63 at.% Cr, although it is not shown in experimental phase diagrams (14). More experimental work using rigorous X-ray diffraction techniques needs to be done to substantiate the L_{12} and L_{10} phase regions. A database for the Pt-Al-Cr-Ni system will be created. Many experiments on the Pt-rich side of the Pt-Al-Cr-Ni system have already been performed (22). The addition of nickel increases the curvature of the γ' solvus, as shown in Figure 3 (23).

Conclusions

The principal aim of the ongoing work at the University of Bayreuth is to describe the platinum-rich side of the Pt-Al-Cr-Ni system. It has been demonstrated that *ab initio* calculations can provide

helpful information on the formation of phases to compensate for sparse experimental data on the alloy system. It is hoped that, in the long term, the Bayreuth database can be merged with the Pt-Al-Cr-Ru database, which is being developed by Mintek and the University of Leeds (see Part II of this series of papers (2)).

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Table I

Key to Figure 3 (EPMA = electron probe micro analyser; SEM-EDX = scanning electron microscopy-energy dispersive X-ray detection)

Symbol	System	Technique	Remarks
●	Pt-Al-Cr-Ni	SEM-EDX	First experimental series
▼	Pt-Al-Cr-Ni	SEM-EDX	Second experimental series
◆	$PtAl_{12}Ni_6$	EPMA	—
○	$PtAl_{12}Cr_6$	EPMA	—
■	$PtAl_{12}Cr_6Ni_5$	SEM-EDX	—
△	Pt-Al-Cr-Ni	SEM-EDX	Further measurements

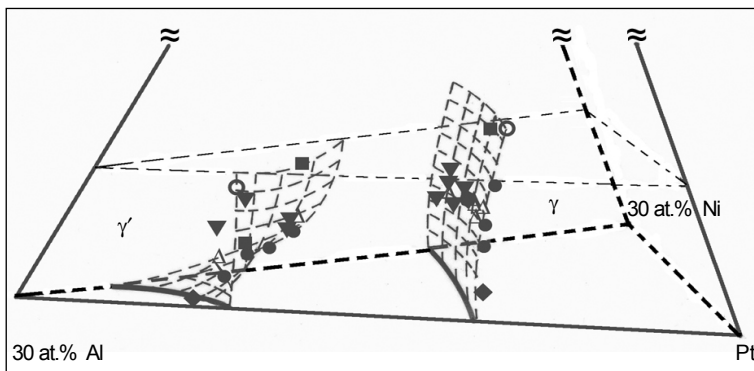


Fig. 3 Phase diagram for the Pt-rich side of the Pt-Al-Cr-Ni system; experimental points (measured by Wenderoth and Vorberg) are taken from Reference (23). See Table I for key

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In his Ph.D. thesis at the University of Bayreuth, Markus Wenderoth developed new precipitation-hardened platinum base alloys for application at high temperatures and experimentally analysed their microstructure, high-temperature strength and oxidation behaviour. He is now working as a production engineer with Siemens Medical Solutions, Vacuum Technology Division.



Professor Dr.-Ing. Uwe Glatzel is head of the Chair of Metals and Alloys at the University of Bayreuth. His work had a big impact on the development of modern high-temperature alloys, mainly nickel base superalloys. He advises several research groups, including those working on platinum-based superalloys and other alloys for high-temperature applications, laser metallurgy, material analysis and artificial knee joints.