

problems of kinetic and morphological peculiarities occurring during hydride phase transformations were reported by Yu. A. Artemenko and co-workers of Donetsk, while allied measurements of changes in the mechanical properties of the palladium/hydrogen system were presented by A. P. Kusin and colleagues, also from the same university.

Further quantitative and theoretical consequences of the effects of hydrogen expansion were discussed. The modelling of hydrogen-induced non-elastic deformations was reported by V. N. Madudin and co-workers, Chelyabinsk, Russia. A paper by Young-sui Cho, University of Kangwon, South Korea, considered the kinetics of hydrogen absorption – determined from measurements of changes in the electrical resistivity of thin palladium films.

The importance of stress/strain gradient behav-

our to bulk lattice hydrogen diffusion processes, as shown by Gorsky and Diffusion Elastic Effect related phenomena, was reported in a paper by R. V. Bucur, Uppsala, Sweden, K. Kandasamy, Jaffna, Sri Lanka, Y. Sakamoto, Nagasaki, Japan, together with X. Q. Tong and F. A. Lewis, Belfast, Northern Ireland, with particular reference to apparent changing trends in the hydrogen diffusion coefficient, D_{H_2} , with the initial values of the hydrogen content.

Lastly, results of hydrogen isotope structure analyses by powder X-ray diffraction, for compositions approximating to $PdH_{0.6}$, $PdD_{0.5}$ and $PdT_{0.66}$ were reported by T. A. Beiter and J. S. Cantrell, of Miami University, Oxford, Ohio.

A number of the papers presented at this Donetsk conference will be published in a special issue of the *International Journal of Hydrogen Energy*.
F.A.L.

Alkane Oxidation by Encapsulated Ruthenium Complexes

The oxidation reaction of alkanes by the usual metal complex catalysts eventually results in deactivation or catalyst degradation. Catalysts, such as porphyrin- or phthalocyanine-based complexes, can acquire improved stability by a halogenation treatment. However, these systems still eventually degrade. Encapsulating catalytic materials within other structures is now a routine operation, and has resulted in many interesting properties (see, for example, *Platinum Metals Rev.*, 1994, **38**, (2), 59) including enhanced catalytic activity. Therefore, isolating the phthalocyanine catalyst, to remove it from outside attack, by encasing it within zeolite pores was tried as a way to improve its lifetime. However, there are still drawbacks caused by the peroxide and iodosylbenzene oxidants.

Recent work has involved encapsulating an iron phthalocyanine, FePc, catalyst in zeolite within a polymer matrix, and this has been successful in producing a more active and stable catalyst. A perfluorinated FePc complex catalyst, however, was as sensitive to the oxidant as FePc.

Based on this researchers at the University of Texas at Dallas, U.S.A., (K. J. Balkus, M. Eissa and R. Levado, *J. Am. Chem. Soc.*, 1995, **117**, (43), 10753–10754) prepared a fluorinated ruthenium phthalocyanine catalyst, $RuF_{16}Pc$, and encapsulated it in zeolite NaX. It was used together with *tert*-butyl hydroperoxide oxidant for the oxidation of cyclohexane and cyclohexanol at room temperature.

Encapsulation prevented dimerisation of the complex and enhanced the activity. Even though the catalyst now contained less metal, it had over ten times the activity of the homogeneous catalyst. Cyclohexane was oxidised primarily to the ketone at a rate of almost 3000 turnovers/day and no deactivation was seen after over 20,000 turnovers. Placing the $RuF_{16}Pc$ into the zeolite increased its selectivity towards cyclohexanone, compared with the homogeneous catalyst. The peroxide efficiency was also improved.

This system may be one of the best low-temperature peroxide-based oxidation catalysts.

Palladium Alloy Mechanical Properties

The mechanical properties of three structural forms of palladium-manganese alloy, containing 10 to 25 atomic per cent manganese, have been examined (D. Jonsen, A. Moss, J. Shenk, K. Rebeiz, S. Nesbit, R. Foley and A. Craft, *Mater. Sci. Eng.*, 1995, **A199**, (2), 131–138). One form, with $L1_2$ structure, produced by the presence of hydrogen during heat treating the alloy, had markedly different properties to the other two forms; these included enhanced strength and hardness, reduced ductility and fatigue life. The differences are thought due to the formation of the $L1_2$ structure and a hydride phase in this alloy whose decomposition significantly affects the properties. The hydrogen-induced lattice rearrangement may be a way of manipulating the mechanical properties of alloys.